	Lecture Plan							
Name of the college: Govt. college of Arts Sci. and Com. Sanquelim								
Name of Faculty: Ms. Varsha K. Sail	Subject:Chemistry							
Paper code: CHC-106	Program: T Y BSc	Division: A						
Academic year: 2024 - 2025	Semester: V	Total Lectures: 15						
Course Objectives: Chemistry of different compounds of halogen, introduction to solid state chemistry,								
Expected Course Outcome: Student gain knowledge about different type of compounds of halogen, the defects in solids and the periodic propertties								
Student Learning Outcome: Students learn and understand the basics of different compounds of halogens, their properties and are introduced to								

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			No. of				
Month	Lecture From	Lecture To	lecture s allotte d per week	Topic, Subtopic to be covered	Exerci se/ Assign ment	ICT Tools	Reference books
June	28/06/24	06/07/24	01	Topic 1 <u>Chemistry of halogens</u> - Prsession – halogen and its properties, Introduction to componds of halogen. Interhalogen classification and types.		Smart board, PPT and chalk and black board	Lee, J.D. Concise Inorganic Chemistry ELBS, 1991. 2. Cotton, F.A., Wilkinson, G. & Gaus, P.L. Basic Inorganic Chemistry, 3rd ed., Wiley. 3. Douglas, B.E., McDaniel, D.H. & Alexander, J.J. Concepts and Models in Inorganic Chemistry, John Wiley & Sons. 4. Huheey, J.E., Keiter, E.A.,

					Keiter, R.L. &
					Medhi, O.K.
					Inorganic
					Chemistry:
					Principles of
					Structure and
					Reactivity,
					Pearson
					Education India,
					2006.
				Smart	
				board,	
				PPT and	
			General methods of preparation and	chalk	
			chemical properties of :of	and	
			Interhalogens, structure, bonding of	black	
July	8/07/24	13/07/24	interhalogen	board	
			iii) Oxyacids of halogens in different		
			oxidation states: General methods of		
			preparation and chemical		
	15/07/04	20/7/24	properties structure, bonding and		
	15/07/24	20/7/24	acid strength		
	22/07/24	27/07/24	11) Polynalide lons General methods of propagation and		
	22/07/24	2/08/24	shamical proporties		
Angust	20/07/24	3/06/24			
August	3/08/24	10/08/24	structure, bonding		
	12/08/24	17/08/24	of preparation and chemical properties		
	10/08/24	24/08/24	structure bonding and acid strongth		
	19/06/24	24/00/24	Topic 2. Inorganic Solid State		
			Chamistry Defacts in		
	26/08/24	31/08/24	solide Thermodynamic basis for		
	20/08/24	51/08/24	sonus, mermouynamic dasis for		

			defect.		
			Types of defect. Point defects;		
Septem			Schottky and Frenkel defects,		
ber	2/09/24	14/09/24	Colour centre,		
	16/09/24	21/09/24	Extended defects and Non- stoichiometry		
			Consequenses of defect		
	23/09/24	28/09/24			
			Band Theory of solids: Band gaps,		
			Metals,		
	30/09/24	05/10/24			
Octobe			, Band gaps Insulators and Semi-		
r	07/10/24	12/10/24	conductors		
	14/10/24	22/10/24	revision		

* Assessment Rubrics

Component	Max Marks
ISA 1	10
ISA 2	
Practical	
Project	
Semester	
End Exam	

Name of Faculty: varsha K sail	Subject: Inorganic chemistry	
Paper code:CHC-106	Program: TY BSc	Division: Batch i and iv
Academic year: 2023- 2024	Semester: V	Total Practicals/Labs: 60 hrs
Credits:02		
Course Objectives: Learn		
estimation of metal by		
gravimetric method and		
synthesis of inorganic		
compounds		
Expected Course Outcome:		
student learn technique of		
gravimetric estimation and		
preparation of compounds,		
understand theory and		
calculation involved.		
Student Learning Outcome:		
Learn technique of		
Gravimetric estimation and		
preparation. Student should		
be able to understand theory		
and do calculation		

Month	Practic Schedu Date	als/Labs Jled	No. of Practicals/La bs planned	List of Experiments	Reference books
June	28/06/2 4	06/07/ 24			 G.H. Jeffery, J. Bassett, J. Mendham, R. C. Denney, Vogel's Textbook of
					Quantitative Chemical Analysis, 5 th Edn. ELBS
				Practical batches/workload not yet allotted	
July	8/07/24	13/07/ 24			
			2 practical session of 4 hours each	To estimate the amount of Al as Al_2O_3 in the given solution of aluminium sulphate.	
	15/07/2 4	20/7/2 4		Brien seranon er mannann sarpiner.	
			2 practical session of 4	To estimate the amount of Fe as Fe_2O_3 in the	
			hours each	given solution of ferric chloride containing barium	
	22/07/2 4	27/07/ 24		chloride and free HCI.	
			2 practical session of 4		2.S. Ratan, Experiments in Applied Chemistry, 3 rd Edn. S.K. Kataria & Sons
			nouis cacil	To estimate the amount of nickel as Ni-DMG in	3.O. P. Pandey, D. N. Bajpai and S. Giri, Practical Chemistry Revised Edn S
	20/07/2	2/09/2		the solution of nickel chloride containing copper	Chand.
	28/07/2 4	3/08/2 4		chloride and free HCl	

August	5/08/24	10/08/ 24	2 practical session of 4 hours each	To estimate the amount of barium as $BaCrO_4$ in the solution of barium chloride containing ferric chloride and free HCl.	
	12/08/2 4	17/08/ 24	2 practical session of 4 hours each	To estimate the amount of Zinc as $Zn_2P_2O_7$ in the given solution of zinc sulphate containing copper sulphate and free H_2SO_4 .	
	19/08/2 4	24/08/ 24	2 practical session of 4 hours each	Preparation Potassium trioxalatoferrate (III).	
	26/08/2 4	31/08/ 24	2 practical session of 4 hours each	Preparation of potassium trioxalatoaluminate(III).	
Septem ber	2/09/24	14/09/ 24	2 practical session of 4 hours each	Preparation of Tristhioureacopper(I) sulphate	
	16/09/2 4	21/09/ 24	2 practical session of 4 hours each	Preparation of Cobalt (III)Hexammine complex	
	23/09/2 4	28/09/ 24		revision	
	30/09/2 4	05/10/ 24		revision	
October	07/10/2 4	12/10/ 24		Exam	
	14/10/2 4	22/10/ 24		Exam	

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