

## Practical Plan

**Name of the college: Government College of Arts, Science & Commerce, Sanquelim, Goa**

**Name of Faculty: Dr. Dattaprasad D. Narulkar**    **Subject: Chemistry**

**Paper code: CHC-203 Inorganic Chemistry - I**    **Program: S.Y.BSc.**    **Division:**

**Academic year: 2025 - 2026**    **Semester: IV**    **Total Practical/Labs: 15 (30 hours)**

**Credits:**

**1**

**Course Objectives:**

- To apply the fundamental theoretical aspects of qualitative inorganic analysis.
- 2. To use various titrimetric techniques to estimate the analytes.
- 3. To use gravimetric methods to estimate metal ions.
- 4. To prepare inorganic coordination compounds.

**Expected Course Outcome:**

At the end of the course students will be able:

**CO1:** perform a qualitative analysis of inorganic mixtures.

**CO2:** prepare coordination compounds of transition elements.

**CO3:** determine unknown concentration of analytes using volumetric and gravimetric procedures.

**Student Learning Outcome:**

At the end of the course students will be able:

**LO1:** acquire knowledge and skill of basic volumetric and gravimetric estimations.

**LO2:** to get hands on experience on qualitative analysis of inorganic mixtures

**LO3:** to get hands on experience on preparation of coordination compounds of transition elements.

<b>Month</b>	<b>Practical/Labs Scheduled Date</b>	<b>No. of Practical /Labs planned</b>	<b>List of Experiments</b>	<b>Reference books</b>
December	2/12/2025	1	Estimate the amount of Ni as bis(dimethylglyoximato) nickel (II) in the given solution of nickel chloride using counter poise method.	Ref 1 and 2
	09/12/2025	1	Estimation of Fe (II) ions by titrating it with $K_2Cr_2O_7$ using the internal indicator.	Ref 1 and 2
	16/12/2025	1	Estimation of the amount of nickel in the given nickel sulphate solution (EDTA method).	Ref 1 and 2
	23/12/2025	1	Preparation of chrome red	Ref 1 and 2.
January	06/12/2025	1	Preparation of tris-(ethylenediamine)nickel (II)chloride	Ref 1 and 2
	13/01/2026	1	Estimation of Mn as manganese pyrophosphate present in the given manganese sulphate solution.	Ref 1 and 2
	20/01/2026	1	Semi-micro qualitative analysis - I	Ref 1 and 2
	27/01/2026	1	Semi-micro qualitative analysis - I	Ref 1 and 2
February	03/02/2026	1	Semi-micro qualitative analysis - II	Ref 1 and 2
	10/02/2026	1	Semi-micro qualitative analysis - II	Ref 1 and 2
	17/02/2026	1	Semi-micro qualitative analysis - III	Ref 1 and 2

	24/02/2026	1	Semi-micro qualitative analysis - III	Ref 1 and 2
March	03/02/2026	1	Semi-micro qualitative analysis - IV	Ref 1 and 2
	10/02/2026	1	Semi-micro qualitative analysis - IV	Ref 1 and 2
	17/02/2026	1 1	Repeat Practical	Ref 1 and 2
	24/03/2026	1	Repeat Practical	Ref 1 and 2
	31/03/2026	1	Practical exam	Ref 1 and 2

### References

1. Vogel's Text book of Qualitative analysis
2. Vogel's Textbook of Quantitative Analysis

* Assessment Rubrics	
Component	Max Marks
ISA	15
Practical	25
Semester End Exam	60
Total	100